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(S) Fluoran compounds.

(57) A fluoran compound represented by general formula:

where each of  $R_1$  and  $R_2$  is a  $C_1$ - $C_4$  alkyl group, a cycloalkyl group or an aryl group, or  $R_1$  and  $R_2$  may form together with N a saturated ring,  $R_3$  is a benzyl or phenyl group which may be substituted, and  $R_4$  is a  $C_1$ - $C_4$  alkyl group.

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# FLUORAN COMPOUNDS

The present invention relates to novel fluoran compounds. More particularly, the invention relates to 2-substituted amino-3-alkoxy-6-di-substituted amino-fluoran compounds which are useful as color precursors for heat sensitive record sheets or electrical heat sensitive

disclosed in Japhanese Examined Patent Publications No. 2406/1973, No.43296/1973, No. 32767/1974 and

No. 23204/1976, and Japanese Unexamined Patent Publication No. 34526/1974. These fluoran compounds are used as color precursors for heat sensitive record sheets or electrical heat sensitive record sheets. However, heat sensitive record sheets wherein these fluoran compounds are used as

record sheets. Certain fluoran compounds have been

15 color precursors, have various drawbacks, for instance, in the developed color density, the initial color density, the color-development initiation temperature and the rising for color-development. Therefore, they cannot provide adequate properties required for heat sensitive

20 record sheets. For instance, 2-anilino-3-methyl-6-

diethylamino-fluoran disclosed in the above-mentioned Japanese Examined Patent Publications, tends to undergo partial color development during the preparation of heat sensitive record sheets, whereby the initial color density of the record sheets tends to be high. On the other hand, with a heat sensitive record sheet wherein 2-(2-chloro-phenylamino)-6-diethylamino-fluoran is used, the initial color density is low, but the color-deveopment initiation temperature is too high and the rising for the color development is inadequate. Further, 2-anilino-3-methyl-6-N-methyl-N-cyclohexylamino-fluoran does not provide an adequate rising for color development.

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The rising for color-development means a rising of a curve in a diagram of color density-color development temperature curve given by plotting color densities on the ordinate and color-development temperatures on the abscissas as a value given by multiplying 100 to  $\tan\theta$  in the maximum slant of the curve.

researches for fluoran compounds to be used for heat sensitive record sheets which have no substantial slef-color development, a high developed color density, a low color-development initiation temperature and a high rising for color-development, and have finally found that 2-substituted amino-3-alkoxy-6-di-substituted amino-fluoran compounds have excellent properties to satisfy the above requirements.

Namely, the present invention provides fluoran compounds represented by general formula:

where each of R<sub>1</sub> and R<sub>2</sub> is a C<sub>1</sub>-C<sub>4</sub> alkyl group, a cycloalkyl group or an aryl group, or R<sub>1</sub> and R<sub>2</sub> may form together with N a saturated ring, R<sub>3</sub> is a benzyl or phenyl group which may be substituted, and R<sub>4</sub> is a C<sub>1</sub>-C<sub>4</sub> alkyl group.

The present invention also provides a heat sensitive record sheet which comprises a coated layer comprising a fluoran compound represented by the above general formula I.

Now, the present invention will be described in detail with reference to the preferred embodiments.

The fluoran compounds represented by the general

formula I according to the present invention are colorless
or slightly colored solids which are stable in air and
which, when brought in contact with an acidic substance,
immediately form coloring agents having a dark black
color. The developed coloring agents have excellent
storage stability and are therefore quite useful.

The fluoran compounds represented by the general formula I according to the present invention may be

prepared by following method (1),(2) or (3).

(1) An aniline derivative represented by the general formula:

where R is a hydrogen atom or a C<sub>1</sub>-C<sub>4</sub> alkyl group, and R<sub>3</sub>

5 and R<sub>4</sub> are as defined above with respect to the general formula I, and a benzophenone compound represented by the general formula:

$$\begin{array}{c}
R_1 \\
R_2
\end{array} > N \longrightarrow C \longrightarrow C$$
[III]

where R<sub>1</sub> and R<sub>2</sub> are as defined above with respect to the general formula I, are reacted in the presence of

10 concentrated sulfuric acid at a temperature of from 0 to 80°C for several hours. After the reaction, the reaction mixture is poured into water and then adjusted to a pH of from 8 to 10 with an addition of sodium hydroxide, whereupon the precipitates are collected by filtration.

15 To the cake thus obtained, toluene and an aqueous solution

To the cake thus obtained, toluene and an aqueous solution containing from 5 to 10% of sodium hydroxide are added, and the mixture is stirred for from 1 to 3 hours under reflux, whereupon the toluene layer is separated by liquid separation, washed with water and then concentrated. The

precipitated crystals are collected by filtration. The crystals are dried, whereby a slightly colored 2-substituted amino-3-alkoxy-6-di-substituted amino-fluoran represented by the general formula I is obtainable in high purity and high yield. If necessary, the product is recrystallized from a volatile organic solvent such as tolene, acetone, butylacetate or hexane.

(2) A 2-amino-3-alkoxy-6-di-substituted amino-fluoran represented by the general formula:

$$\begin{array}{c} R_1 \\ R_2 \\ \end{array}$$

$$\begin{array}{c} O \\ O \\ \end{array}$$

$$\begin{array}{c} O \\ N \\ H_2 \\ \end{array}$$

$$[IV]$$

- where R<sub>1</sub>, R<sub>2</sub> and R<sub>4</sub> are defined above with respect to the general formula I, is reacted with a substituted benzyl chloride or a substituted benzyl bromide in a volatile organic inert solvent in the presence of an acid-binding agent such as sodium bicarbonate, or with a substituted bromobenzene or chlorobenzene in a volatile organic inert solvent in the presence of potassium carbonate by using copper powder and iodine as catalysts, whereby a 2-substituted amino-3-alkoxy-6-di-substituted amino-fluoran represented by the general formula I is obtainable.
- 20 (3) A 2-substituted amino-3-hydoroxy-6-di-substituted amino-fluoran represented by the general formula:

where R<sub>1</sub>, R<sub>2</sub> and R<sub>3</sub> are as defined above with respect to the general formula I, is reacted with an alkylating agen such as dimethyl sulfate, methyl p-toluene sulfonate or methyl iodide in the presence of an alkali such as sodium by hydroxide in water or in a volatile organic inert solvent whereby a 2-substituted amino-3-alkoxy-6-di-substituted amino-fluoran represented by the general formula I is obtainable.

From the practical point of view, the method (1) is 10 preferred.

As representative compounds of the above general formula I of the present invention, there may be mentioned 2-benzylamino-3-methoxy-6-diethylamino-fluoran, 2-benzylamino-3-methoxy-6-di-n-butylamino-fluoran, 2-benzylamino-3-methoxy-6-N-methyl-N-cyclohexylamino-fluoran, 2-anilino-3-methoxy-6-diethylamino-fluoran, 2-anilino-3-methoxy-6-diethylamino-fluoran, 2-(2-methylphenylamino)-3-methoxy-6-diethylamino-fluoran, 2-(2-methylphenylamino)-3-methoxy-6-diethylamino-fluoran, 2-(4-methylphenylamino)-3-methoxy-6-diethylamino-fluoran, 2-(4-methylphenylamino)-3-methoxy-6-diethylamino-fluoran, 2-(4-methylphenylamino)-3-methoxy-6-diethylamino-fluoran, 2-(4-methylphenylamino)-3-methoxy-6-di-n-butylamino-

fluoran, 2-(2-chlorophenylamino)-3-methoxy-6-diethyl-

amino-fluoran, 2-(2-chlorophenylamino)-3-methoxy-6-di-nbuthylamino-fluoran, 2-(4-chlorophenylamino)-3-methoxy-6diethylamino-fluoran, 2-(4-chlorophenylamino)-3-methoxy-6di-n-butylamino-fluoran, 2-anilino-3-ethoxy-6-diethyl-5 amino-fluoran, 2-anilino-3-ethoxy-6-di-n-butylaminofluoran, 2-anilino-3-ethoxy-6-N-methyl-N-cyclohexylaminofluoran, 2-anilino-3-ethoxy-6-N-ethyl-N-p-tolylaminofluoran, 2-(2-methylphenylamino)-3-ethoxy-6-diethylaminofluoran, 2-(2-methylphenylamino)-3-ethoxy-6-di-n-butylamino-fluoran, 2-(4-methylphenylamino)-3-ethoxy-6-diethylamino-fluoran, 2-(4-methyphenylamino)-3-ethoxy-6-di-nbutylamino-fluoran, 2-(4-methylphenylamino)-3-ethoxy-6-Nethyl-N-p-tolylamino-fluoran, 2-(4-chlorophenylamino)-3ethoxy-6-diethylamino-fluoran, 2-(4-chlorophenylamino)-3ethoxy-6-di-n-butylamino-fluoran, 2-(4-chlorophenylamino)-3-ethoxy-6-N-ethyl-N-p-tolylamino-fluoran, 2-anilino-3methoxy-6-pyrrolidinyl-fluoran, 2-anilino-3-methoxy-6-N-ethyl-N-cyclohexylamino-fluoran, 2-anilino-3-propoxy-6-diethylamino-fluoran, 2-anilino-3-propoxy-6-di-n-20 butylamino-fluoran, 2-anilino-3-butoxy-6-diethylaminofluoran, 2-anilino-3-butoxy-6-di-n-butylamino-fluoran.

As typical representatives of the aniline derivative represented by the general formula II to be used in the present invention, there may be mentioned N-benzyl-2,4-25 dimethoxyaniline, N-phenyl-2,4-dimethoxyaniline, N-(2-methylphenyl)-2,4-dimethoxyaniline, N-(4-methylphenyl)-2,4-dimethoxyaniline, N-(2-chlorophenyl)-2,4-dimethoxy-

aniline, N-(4-chlorophenyl)-2,4-dimethoxyaniline, N-benzyl-4-hydroxy-2-methoxyaniline, N-phenyl-4-hydroxy-2-methoxyaniline, N-phenyl-4-hydroxy-2-methoxyaniline, N-(2-methyl-phenyl)-2,4-diethoxyaniline, N-(4-methylphenyl)-2,4-

- diethoxyaniline, N-(2-chlorophenyl)-2,4-diethoxyaniline, N-(4-chlorophenyl)-2,4-diethoxyaniline and N-phenyl-4hydroxy-2-ethoxyaniline. Among the above aniline derivatives, those substituted by an alkoxy group at 4-position are preferred from the practical point of view.
- As the benzophenone compound represented by the above general formula III to be used in the present invention, there may be mentioned 2-hydroxy-4-dimethylamino-2'-carboxy-benzophenone, 2-hydroxy-4-diethylamino-2'-carboxy-benzophenone, 2-hydroxy-4-dipropylamino-2'-carboxy-
- benzophenone, 2-hydroxy-4-dibutylamino-2'-carboxybenzophenone, 2-hydroxy-4-N-methyl-N-cyclohexylamino-2'carboxybenzophenone, 2-hydroxy-4-N-ethyl-N-p-tolylamino-2'carboxy-benzophenone, 2-hydroxy-4-pyrrolidinyl-2'-carboxybenzophenone and 2-hydroxy-4-piperidino-2'-carboxy-benzophenone.

As representatives of the 2-amino-3-alkoxy-6-di-substituted amino-fluoran represented by the above general formula IV to be used in the present invention, there may be mentioned 2-amino-3-methoxy-6-dimethylamino-fluoran,

25 2-amino-3-methoxy-6-diethylamino-fluoran, 2-amino-3-methoxy-6-dipropylamino-fluoran, 2-amino-3-methoxy-6-dibutylamimo-fluoran, 2-amino-3-methoxy-6-N-methyl-N-

cyclohexylamino-fluoran, 2-amino-3-methoxy-6-N-ethyl-N-p-tolylamino-fluoran, 2-amino-3-methoxy-6-pyrrolidinyl-fluoran, 2-amino-3-methoxy-6-piperidino-fluoran, 2-amino-3-ethoxy-6-dimethylamino-fluoran, 2-amino-3-ethoxy-6-dipropylamino-fluoran, 2-amino-3-propoxy-6-dipropylamino-fluoran, 2-amino-3-propoxy-6-dibutylamino-fluoran, 2-amino-3-butoxy-6-N-methyl-N-cyclohexylamino-fluoran and 2-amino-3-butoxy-6-N-ethyl-N-p-tolylamino-fluoran.

As representatives of the 2-substituted amino-3
10 hydroxy-6-di-substituted amino-fluoran represented by the above general formula V to be used in the present invention, there may be mentioned 2-benzylamino-3-hydroxy-6-diethylamino-fluoran, 2-anilino-3-hydroxy-6-diethyl
15 amino-fluoran, 2-anilino-3-hydroxy-6-diethyl
15 amino-fluoran, 2-anilino-3-hydroxy-6-dipropylaminofluoran, 2-anilino-3-hydroxy-6-dibutylamino-fluoran,

2-anilino-3-hydroxy-6-N-methyl-N-cyclohexylamino-fluoran,

2-anilino-3-hydroxy-6-N-ethyl-N-p-tolylamino-fluoran,

2-(4-methylphenylamino)-3-hydroxy-6-diethylamino-fluoran,

2-(2-methylphenylamino)-3-hydroxy-6-diethylamino-fluoran,

2-(4-chlorophenylamino)-3-hydroxy-6-diethylamino-fluoran

and 2-(2-chlorophenylamino)-3-hydroxy-6-diethylamino-fluoran

As the substituted benzylchloride or the substituted 25 benzylbromide to be used in the present invention, there may be mentioned benzylchloride, benzylbromide, 2-chlorobenzylchloride, 3-chlorobenzyl-

chloride, 4-chlorobenzylchloride, 2-methylbenzylchloride, 4-methylbenzylchloride, 3-methylbenzylchloride and 2-methylbenzylbromide.

As the substituted bromobenzene or substituted

5 chlorobenzene to be used in the present invention, there
may be mentioned 2-bromochlorobenzene, o-dichlorobenzene, 3-bromochlorobenzene, 4-bromochlorobenzene,
bromobenzene, 2-bromotoluene, 4-bromotoluene or 3-bromotoluene.

invention, there may be mentioned concentrated sulfuric acid, acetic acid anhydride, phosphoric acid, polyphosphoric acid, phosphorus oxychloride and zinc chloride. From the practical point of view, it is preferred to use 15 concentrated sulfuric acid which serves as a solvent for a benzophenone compound represented by the above general formula III and at the same time serves as a condensing agent.

Now, the present invention will be described in 20 further detail with reference to Examples for the preparation of typical 2-substituted amino-3-alkoxy-6-disubstituted amino-fluoran compounds of the present invention and Application Examples.

### Example 1

25 2-Benzylamino-3-methoxy-6-diethylamino-fluoran

(Compound No. A)

To 150 g of 95% sulfuric acid, 12.5 g of 2-hydroxy-

4-diethylamino-2'-carboxy-benzophenone was added and completely dissolved at a temperature of about 20°C, and then 9.7 g of N-benzyl-2,4-dimethoxyaniline was added and reacted therewith at a temperature of from 10 to 20°C for 5 24 hours. After the reaction, the reaction mixture was poured into 1 liter of ice water and then adjusted to a pH of from 7 to 8 by an addition of 10% sodium hydroxide aqueous solution, whereupon the precipitates were collected by filtration. To the cake thus obtained, 300 ml of toluene and 150 ml of a 10% sodium hydroxide aqueous solution were added, and the mixture was stirred for 2 hours under reflux. Then, the toluene layer was separated by liquid separation, then washed with water, concentrated to dryness and solidified by an addition of hexane. The solid thereby obtained was dried to obtain 20.5 g of slightly pink 2-benzylamino-3-methoxy-6-diethylaminofluoran. The melting point of this product was from 72 to 76°C. Further, this product had  $\lambda_{max}$  at 439 nm (a molecular extinction coefficient (hereinafter referred to simply as "MEC"):  $1.42 \times 10^4$ ) and at 583 nm (MEC: 3.91  $\times$ 20 104) as measured in 95% acetic acid. A solution of this product in toluene was colorless. When brought in contact with silica gel, the product readily underwent color-development and turned purplish black.

The N-benzyl-2,4-dimethoxyaniline used as a starting material was prepared in the following manner.

Into 150 ml of ethanol, 30.6 g of 2,4-dimethoxy-

aniline and 21.2 g of benzaldehyde were dissolved, and 2 g of acetic acid was added. The mixture was reacted at a temperature of from 50 to 60°C for 4 hours. The reaction solution was poured into 500 ml of water and extracted 5 with 300 ml of toluene. The toluene solution was separated by liquid separation, then washed with water and concentrated to dryness. The oily product thereby obtained was dissolved into 150 ml of ethanol, and 11.3 g of sodium boro-hydride was added in 2 hours at a temperature of from 30 to 40°C and reacted therewith. 10 Further, the mixture was stirred for one hour at a temperature of 40°C. The reaction mixture was poured into 1 liter of water and extracted with 500 ml of toluene. The toluene layer was washed with water, concentrated and 15 purified by distillation under reduced pressure, whereby 40.9 g of N-benzyl-2,4-dimethoxyaniline was obtained. product was a liquid having a boiling point of from 166 to 168°C/2 mmHq.

#### Exampl 2

25

20 2-Benzylamino-3-methoxy-6-di-n-butylamino-fluoran (Compound No. B)

To 70 g of 95% sulfuric acid, 7.4 g of 2-hydroxy-4-di-n-butylamino-2'-carboxy-benzophenone and 4.9 g of N-benzyl-2,4-dimethoxyaniline were added and reacted in the same manner as in Example 1, whereby 9.8 g of slightly purple 2-benzylamino-3-methoxy-6-di-n-butylamino-fluoran was obtained. The melting point of this product was from

134 to 136°C. The product had λ max at 443 nm (MEC: 1.59 x 10<sup>4</sup>) and at 587 nm (MEC: 3.95 x 10<sup>4</sup>) as measured in 95% acetic acid. A solution of this product in toluene was colorless. When brought in contact with silica gel, the product readily underwent color-development and turned blackish purple.

# Example 3

2-Benzylamino-3-methoxy-6-N-methyl-N-cyclohexylamino-fluoran (Compound No. C)

10 To 70 g of 95% sulfuric acid, 7.1 g of 2-hydroxy-4-N-methyl-N-cycolhexylamino-2'-carboxy-benzophenone and 4.9 g of N-benzyl-2,4-dimethoxyaniline were added and reacted in the same manner as in Example 1. The product was recrystallized from toluene, whereby 8.3 g of slightly 15 purple 2-benzyl-3-methoxy-6-N-methyl-N-cyclohexylamino-fluoran was obtained. The melting point of this product was from 210 to 211.5°C. The product had λmax at 443 nm (MEC: 1.55 x 10<sup>4</sup>) and at 587 nm (MEC: 3.94 x 10<sup>4</sup>) as measured in 95% acetic acid. A solution of this product in toluene was colorless. When brought in contact with silica gel, the product readily underwent color-development and turned blackish purple.

#### Example 4

25

2-Anilino-3-methoxy-6-diethylamino-fluoran (Compound No. D)

To 150 g of 95% sulfuric acid, 15.7 g of 2-hydroxy-4-diethylamino-2'-carboxy-benzophenone and 11.5 g of N-

phenyl-2,4-dimethoxy aniline were added and reacted in the same manner as in Example 1. The product was recrystallized from toluene, whereby 13.0 g of white 2-anilino-3-methoxy-6-diethylamino-fluoran was botained.

The melting point of this product was from 104 to  $104.5^{\circ}$ C. The product had  $\lambda_{\text{max}}$  at 449 nm (MEC: 1.46 x  $10^4$ ) and at 591 nm (MEC: 3.11 x  $10^4$ ) as measured in 95% acetic acid. A solution of this product in toluene was colorless. When brought in contact with silica gel, the product readily underwent color-development and turned bluish black.

The N-phenyl-2,4-dimethoxyaniline used as starting material was prepared in the following manner.

To a mixture of 58.5 g of N-acetyl-2,4-dimethoxyaniline and 70.7 g of bromobenzene, 29 g of anhydrous potassium carbonate, 2.1 g of copper powder and 0.8 g of iodine were added and reacted at a temperature of from 170 to 180°C for 20 hours. Then, to this reaction mixture, 30 g of potassium hydroxide and 100 ml of isoamyl alcohol were added and reacted at a temperature of from 120 to 130°C for 3 hours. To this reaction mixture, 300 ml of 20 toluene was added, and the toluene layer was separated by liquid separation, washed with water, concentrated and then purified by distillation under reduced pressure, whereby 40 g of oily N-phenyl-2,4-dimethoxyaniline was obtained. The boiling point of this product was from 161 25 to  $164^{\circ}$ C/2 mmHg.

### Example 5

2-Anilino-3-methoxy-6-di-n-butylamino-fluoran (Compound No. E)

To 150 g of 95% sulfuric acid, 18.5 g of 2-hydroxy
4-di-n-butylamino-2'-carboxy-benzophenone and 11.5 g of N-phenyl-2,4-dimethoxyaniline were added and reacted in the same manner as in Example 1. The product was recrystallized from toluene, whereby 20.9 g of white 2-anilino-3-methoxy-6-di-n-butylamino-fluoran was obtained.

The melting point of this product was from 169 to 172°C. The product had \(\lambda\_{max}\) at 45l nm (MEC: 1.52 x 10<sup>4</sup>) and at 593 nm (MEC: 3.29 x 10<sup>4</sup>) as measured in 95% acetic acid. A solution of this product in toluene was colorless. When brought in contact with silica gel, the product readily underwent color-development and turned bluish black.

# Example 6

2-Anilino-3-methoxy-6-N-methyl-N-cyclohexylamino-fluoran (Compound No. F)

To 60 g of 95% sulfuric acid, 7.1 g of 2-hydroxy-4
20 N-methyl-N-cyclohexylamino-2'-carboxybenzophenone and 4.6 g of N-phenyl-2,4-dimethoxyaniline were added and reacted in the same manner as in Example 1. The product was recrystallized from acetone, whereby 4.4 g of white 2-anilino-3-methoxy-6-N-methyl-N-cyclohexylamino-fluoran

25 was obtained. The melting point of this product was from 201 to 203.5°C. The product had λ<sub>max</sub> at 450 nm (MEC: 1.63 x 10<sup>4</sup>) and at 593 nm (MEC: 3.45 x 10<sup>4</sup>) as measured in 95%

acetic acid. A solution of this product in toluene was colorless. When brought in contact with silica gel, the product readily underwent color-development and turned bluish black.

#### 5 Example 7

2-Anilino-3-methoxy-6-N-ethyl-N-p-tolylamino-fluoran (Compound No. G)

To 60 g of 95% sulfuric acid, 7.5 g of 2-hydroxy-4-Nethyl-N-p-tolylamino-2'-carboxy-benzophenone and 4.6 g of 10 N-phenyl-2,4-dimethoxyaniline were added and reacted in the same manner as in Example 1. The product was recrystallized from toluene, whereby 6.8 g of white 2anilino-3-methoxy-6-N-ethyl-N-p-tolylamino-fluoran was The melting point of this product was from 196 obtained. The product had  $\lambda_{max}$  at 453 nm (MEC: 1.82 x 15 to 199°C.  $10^4$ ) and at 595 nm (MEC: 3.19 x  $10^4$ ) as measured in 95% acetic acid. A solution of this product in toluene was colorless. When brought in contact with silica gel, the product readily underwent color-development and turned bluish black. 20

#### Example 8

2-(2-Metylphenylamino)-3-methoxy-6-diethylaminofluoran (Compound No. H)

To 150 g of 95% sulfuric acid, 15.7 g of 2-hydroxy-425 diethylamino-2'-carboxy-benzophenone and 12.2 g of N-(2methylphenyl)-2,4-dimethoxyaniline were added and reacted
in the same manner as in Example 1. The product was

recrystallized from toluene, whereby 18.7 g of white 2-(2-methylanilino)-3-methoxy-6-diethylamino-fluoran was obtained. The melting point of this product was from 124 to  $127^{\circ}$ C. The product had  $\lambda_{\rm max}$  at 441 nm (MEC: 1.40 x  $10^4$ ) and at 586 nm (MEC: 3.18 x  $10^4$ ) as measured in 95% acetic acid. A solution of this product in toluene was colorless. When brought in contact with silica gel, the product readily underwent color-development and turned purplish black.

The N-(2-methylphenyl)-2,4-dimethoxyaniline used as a starting material was prepared in the same manner as in Example 4 by using N-acetyl-2,4-dimethoxyaniline and 2-bromotoluene as the starting materials. The boiling point of the N-(2-methylphenyl)-2,4-dimethoxyanililne was from 172 to 175°C/2 mmHg.

### Example 9

2-(2-Metylphenylamino)-3-methoxy-6-di-n-butylamino-fluoran (Compound No. I)

To 150 g of 95% sulfuric acid, 18.5 g of 2-hydroxy4-di-n-butylamino-2'-carboxy-benzophenone and 12.2 g of
N-(2-methylphenyl)-2,4-dimethoxyaniline were added and
reacted in the same manner as in Example 1. The product
was recrystallized from toluene, whereby 18.3 g of white
2-(2-methylphenylamino)-3-methoxy-6-di-n-butylamino-fluoran
was obtained. The melting point of this product was from
139 to 142°C. The product had λ<sub>max</sub> at 444 nm (MEC: 1.57 x
10<sup>4</sup>) and at 588 nm (MEC: 3.64 x 10<sup>4</sup>) as measured in 95%

acetic acid. A solution of this product in toluene was colorless. When brought in contact with silica gel, the product readily underwent color-development and turned purplish black.

5 Example 10

2-(4-Metylphenylamino)-3-methoxy-6-diethylaminofluoran (Compound No. J)

To 150 g of 95% sulfuric acid, 15.7 g of 2-hydroxy-4-diethylamino-2'-carboxy-benzophenone and 12.2 g of N-(4-methylphenyl)-2,4-dimethoxyaniline were added and reacted in the same manner as in Example 1. The product was recrystallized from acetone, whereby 22.3 g of white 2-(4-methylphenylamino)-3-methoxy-6-diethylamino-fluoran was obtained. The melting point of this product was from 15 141 to 145°C. The product had λ<sub>max</sub> at 453 nm (MEC: 1.40 x 10<sup>4</sup>) and at 595 nm (MEC: 2.58 x 10<sup>4</sup>) as measured in 95% acetic acid. A solution of this product in toluene was colorless. When brought in contact with silica gel, the product readily underwent color-development and turned 20 black.

The N-(4-methylpnenyl)-2,4-dimethoxyaniline used as a starting material was prepared in the same manner as in Example 4 by using N-acetyl-2,4-dimethoxyaniline and 4-bromotoluene as the starting materials. The boiling point of the N-(4-methylphenyl)-2,4-dimethoxyaniline was from 173 to  $175^{\circ}$ C/2 mmHg.

### Example 11

2-(4-Methylphenylamino)-3-methoxy-6-di-n-butylamino-fluoran (Compound No. K)

To 150 g of 95% sulfuric acid, 18.5 g of 2-hydroxy-5 4-di-n-butylamino-2'-carboxy-benzophenone and 12.2 g of N-(4-methylphenyl)-2,4-dimethoxyaniline were added and reacted in the same manner as in Example 1. The product was recrystallized from toluene, whereby 15.6 g of white 2-(4-methylphenylamino)-3-methoxy-6-di-n-butylamino-fluoran was obtained. 10 The melting point of this product was from 106 to 109°C. The product had  $\lambda_{\text{max}}$  at 455 nm (MEC: 1.65 x  $10^4$ ) and at 600 nm (MEC: 3.14 x  $10^4$ ) as measured in 95% acetic acid. A solution of this product in toluene was colorless. When brought in contact with silica gel, the 15 product readily underwent color-development and turned bluish black.

#### Example 12

2-(2-Chlorophenylamino)-3-methoxy-6-diethylamino-fluoran (Compound No. L)

To 150 g of 95% sulfuric acid, 15.7 g of 2-hydroxy-4-diethylamino-2'-carboxy-benzophenone and 13.2 g of N-(2-chlorophenyl)-2,4-dimethoxyaniline were added and reacted in the same manner as in Example 1. The product was recrystallized from toluene, whereby 13.4 g of white 2-(4-chlorophenylamino)-3-methoxy-6-diethylamino-fluoran was obtained. The melting point of this product was from 121 to 125°C. The product had λ<sub>max</sub> at 437 nm (MEC: 1.29 x

10<sup>4</sup>) and at 577 nm (MEC: 3.63 x 10<sup>4</sup>) as measured in 95% acetic acid. A solution of this product in toluene was colorless. When brought in contact with silica gel, the product readily underwent color-development and turned dark red.

The N-(2-chloropnenyl)-2,4-dimethoxyaniline used as a starting material was prepared in the same manner as in Example 4 by using N-acetyl-2,4-dimethoxyaniline and 2-bromochlorobenzene as the starting materials. The N-(2-chlorophenyl)-2,4-dimethoxyaniline was an oil having a boiling point of from 170 to 177°C/2 mmHg.

Example 13

2-(2-Chlorophenylamino)-3-methoxy-6-di-n-butylamino-fluoran (Compound No. M)

To 125 g of 95% sulfuric acid, 14.8 g of 2-hydroxy-15 4-di-n-butylamino-2'-carboxy-benzophenone and 10.6 g of N-(2-chlorophenyl)-2,4-dimethoxyaniline were added and reacted in the same manner as in Example 1. The product was recrystallized from toluene, whereby 12.2 g of white 2-(2-chlorophenylamino)-3-methoxy-6-di-n-butylamino-fluoran 20 was obtained. The melting point of this product was from The product had  $\lambda_{max}$  at 440 nm (MEC: 1.30 x 105 to 107°C.  $10^4$ ) and at 580 nm (MEC: 3.78 x  $10^4$ ) as measured in 95% A solution of this product in toluene was acetic acid. colorless. When brought in contact with silica gel, the 25 product readily underwent color-development and turned dark red.

# Example 14

2-(4-Chlorophenylamino)-3-methoxy-6-diethylaminofluoran (Compound No. N)

To 150 g of 95% sulfuric acid, 15.7 g of 2-hydroxy-4-diethylamino-2'-carboxy-benzophenone and 13.2 g of N-(4-chlorophenyl)-2,4-dimethoxyaniline were added and reacted in the same manner as in Example 1. The product was recrystallized from toluene, whereby 20.7 g of white 2-(4-chlorophenylamino)-3-methoxy-6-diethylamino-fluoran 10 The melting point of this product was from was obtained. 123 to 127°C. The product had  $\lambda_{\mbox{\scriptsize max}}$  at 449 nm (MEC: 1.34 x $10^4$ ) and at 587 nm (MEC: 2.95 x  $10^4$ ) as measured in 95% acetic acid. A solution of this product in toluene was colorless. When brought in contact with silica gel, the product readily underwent color-development and turned 15 purplish black.

The N-(4-chlorophenyl)-2,4-dimethoxyaniline used as astarting material was prepared in the same manner as in Example 4 by using N-acetyl-2,4-dimethoxyaniline and 4-bromochlorobenzene as the starting materials. 20 N-(4-chlorophenyl)-2,4-dimethoxyaniline was an oil having a boiling point of from 178 to 180°C/2 mmHg. Example 15

2-(4-Chlorophenylamino)-3-methoxy-6-di-n-butylamino-25 fluoran (Compound No. 0)

To 150 g of 95% sulfuric acid, 18.5 g of 2-hydroxy-4-di-n-butylamino-2'-carboxy-benzophenone and 13.2 g of

N-(4-chlorophenyl)-2,4-dimethoxyaniline were added and reacted in the same manner as in Example 1. The product was recrystallized from toluene, whereby 18.0 g of white 2-(4-chlorophenylamino)-3-methoxy-6-di-n-butylamino-fluoran was obtained. The melting point of this product was from 91 to 95°C. The product had λ<sub>max</sub> at 449 nm (MEC: 1.46 x 10<sup>4</sup>) and at 590 nm (MEC: 3.36 x 10<sup>4</sup>) as measured in 95% acetic acid. A solution of this product in toluene was colorless. When brought in contact with silica gel, the product readily underwent color-development and turned bluish black.

Example 16

2-Anilino-3-ethoxy-6-diethylamino-fluoran (Compound No. P)

To 100 g of 95% sulfuric acid, 9.4 g of 2-hydroxy-15 4-diethylamino-2'-carboxy-benzophenone was added and completely dissolved at a temperature of about 20°C, and then 7.7 g of N-phenyl-2,4-diethoxyaniline was added and reacted therewith at a temperature from 10 to 20°C for 24 hours. After the reaction, the reaction mixture was 20 poured into 700 ml of ice water and adjusted to pH of from 7 to 8 by an addition of a 10% sodium hydroxide aqueous solution, whereupon the precipitates were collected by filtration. To the cake thereby obtained, 300 ml of toluene and 150 ml of a 10% sodium hydroxide aqueous 25 solution were added, and the mixture was stirred for 2 hours under reflux. Then, the toluene layer was separated by liquid separation, washed with water, concentrated to dryness and solidified by an addition of hexane. The solidified product was dried to obtain 12.1 g of slightly pink 2-anilino-3-ethoxy-6-diethylamino-fluoran. The melting point of this product was from 84 to 88°C. The product had  $\lambda_{\rm max}$  at 452 nm (MEC: 1.44 x 10<sup>4</sup>) and at 593 nm (MEC: 2.93 x 10<sup>4</sup>) as measured in 95% acetic acid. A solution of this product in toluene was colorless. When brought in contact in with silica gel, the product readily underwent color-development and turned bluish black.

The N-phenyl-2,4-diethoxyaniline used as starting material was prepared in the following manner.

To a mixture of 66.9 g of N-acetyl-2,4-diethoxy-aniline and 70.7 g of bromobenzene, 29 g of anhydrous

15 potassium carbonate, 2.1 g of copper powder and 0.8 g of iodine were added, and the mixture was reacted at a temperature of from 170 to 180°C for 20 hours. Then, to this reaction mixture, 30 g of potassium hydroxide and 100 ml of isoamylalcohol were added, and the mixture was

20 reacted at a temperture from 120 to 130°C for 3 hours. Then, 300 ml of toluene was added thereto. The toluene layer was washed with water, separated by liquid separation, concentrated and then purified by distillation under reduced pressure, whereby 54 g of oily N-phenyl-2,4-di-ethoxyaniline was obtained. The boiling point of this product was from 180 to 185°C/3 mmHq.

Example 17

2-Anilino-3-ethoxy-6-di-n-butylamino-fluoran (Compound No. Q)

To 100 g of 95% sulfuric acid, 11.1 g of 2-hydroxy4-di-n-butylamino-2'-carboxy-benzophenone and 7.7 g of
N-phenyl-2,4-diethoxyaniline were added and reacted in the
same manner as in Example 16. The product was recrystallized from cyclohexene, whereby 11.8 g of white 2-anilino3-ethoxy-6-di-n-butylamino-fluoran was obtained. The
melting point of this product was from 125 to 128°C. The
product had λ<sub>max</sub> at 453 nm (MEC: 1.48 x 10<sup>4</sup>) and at 597 nm
(MEC: 3.17 x 10<sup>4</sup>) as measured in 95% acetic acid. A
solution of this product in toluene was colorless. When
brought in contact with silica gel, the product readily
underwent color-development and turned bluish black.
Example 18

2-Anilino-3-ethoxy-6-N-methyl-N-cyclohexylamino-fluoran (Compound No. R)

To 100 g of 95% sulfuric acid, 10.6 g of 2-hydroxy4-N-methyl-N-cyclohexylamino-2'-carboxybenzophenone and
7.7 g of N-phenyl-2,4-diethoxyaniline were added and
reacted in the same manner as in Example 16. The product
was recrystallized from n-octylchloride, whereby 9.0 g of
white 2-anilino-3-ethoxy-6-N-methyl-N-cyclohexylaminofluoran was obtained. The melting point of this product
was from 169 to 173°C. The product had λ<sub>max</sub> at 452 nm
(MEC: 1.47 x 10<sup>4</sup>) and at 597 nm (MEC: 3.16 x 10<sup>4</sup>) as

measured in 95% acetic acid. A solution of this product in toluene was colorless. When brought in contact with silica gel, the product readily underwent colordevelopment and turned bluish black.

#### 5 Example 19

2-Anilino-3-ethoxy-6-N-ethyl-N-p-tolylamino-fluoran (Compound No. S)

To 100 g of 95% sulfuric acid, 11.3 g of 2-hydroxy-4-N-ethyl-N-p-tolylamino-2'-carboxy-benzophenone and 7.7 g of N-phenyl-2,4-diethoxyaniline were added and reacted in 10 the same manner as in Example 16. The product was recrystallized from n-butylchloride, whereby 12.7 g of white 2-anilino-3-ethoxy-6-N-ethyl-N-p-tolylamino-fluoran The melting point of this product was from was obtained. 15 148 to 152°C. The product had  $\lambda_{\text{max}}$  at 453 nm (MEC: 1.92 x  $10^4$ ) and at 595 nm (MEC: 3.42 x  $10^4$ ) as measured in 95% acetic acid. A solution of this product in toluene was colorless. When brought in contact with silica gel, the product readily underwent color-development and turned 20 bluish black.

#### Example 20

2-(2-Methylphenylamino)-3-ethoxy-6-diethylamino-fluoran (Compound No. T)

To 75 g of 95% sulfuric acid, 6.9 g of 2-hydroxy-4-25 diethylamino-2'-carboxy-benzophenone and 5.4 g of N-(2methylphenyl)-2,4-diethoxyaniline were added and reacted in the same manner as in Example 16. The product was recrystallized, whereby 7.8 g of white 2-(2-methylphenyl-amino)-3-ethoxy-6-diethylamino-flucran was obtained. The melting point of this product was from 197.5 to  $200^{\circ}$ C. The product had  $\lambda_{\text{max}}$  at 443 nm (MEC: 1.46 x  $10^{4}$ ) and at 586 nm (MEC: 3.46 x  $10^{4}$ ) as measured in 95% acetic acid. A solution of this product in toluene was colorless. When brought in contact with silica gel, the product readily underwent color-development and turned purplish black.

The N-(2-methylphenyl)-2,4-diethoxyaniline used as a starting material was prepared in the same manner as in Example 16 by using N-acetyl-2,4-diethoxyaniline and 2-bromotoluene as the starting materials. The boiling point of the N-(2-methylphenyl)-2,4-diethoxyaniline was from 170 to 172°C/0.5 mmHg.

### 15 Example 21

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2-(2-Methyphenylamino)-3-ethoxy-6-di-n-butylaminofluoran (Compound No. U)

To 75 g of 95% sulfuric acid, 8.1 g of 2-hydroxy-4-di-n-butylamino-2'-carboxy-benzophenone and 5.4 g of N-(2-methylphenyl)-2,4-diethoxyaniline were added and reacted in the same manner as in Example 16. The product was recrystallized, whereby 8.1 g of white 2-(2-methyl-phenylamino)-3-ethoxy-6-di-n-butylamino-fluoran was obtained. The melting point of this product was from 125 to  $130^{\circ}$ C. The product had  $\lambda_{max}$  at 444 nm (MEC: 1.49 x  $10^{4}$ ) and at 589 nm (MEC: 3.63 x  $10^{4}$ ) as measured in 95% acetic acid. A solution of this product in toluene was

colorless. When brought in contact with silica gel, the product readily underwent color-development and turned purplish black.

Example 22

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5 2-(4-Methylphenylamino)-3-ethoxy-6-diethylaminofluoran (Compound No. V)

To 100 g of 95% sulfuric acid, 10.3 g of 2-hydroxy-4-diethylamino-2'-carboxy-benzophenone and 8.1 g of N-(4-methtylphenyl)-2,4-diethoxyaniline were added and reacted in the same manner as in Example 16. The product was recrystallized, whereby 11.7 g of white 2-(4-methylphenyl-amino)-3-ethoxy-6-diethylamino-fluoran was obtained. The melting point of this product was from 110 to 113°C. The product had  $\lambda_{\rm max}$  at 454 nm (MEC: 1.47 x 10<sup>4</sup>) and at 597 nm (MEC: 2.85 x 10<sup>4</sup>) as measured in 95% acetic acid. A solution of this product in toluene was colorless. When brought in contact with silica gel, the product readily underwent color-development and turned purplish black.

The N-(4-methylphenyl)-2,4-diethoxyaniline used as a starting material was prepared in the same manner as in Example 16 by using N-acetyl-2,4-diethoxyaniline and 4-bromotoluene as the starting materials. The boiling point of the N-(4-methylphenyl)-2,4-diethoxyaniline was from 195 to 197°C/3 mmHg.

25 Example 23

2-(4-Methylphenylamino)-3-ethoxy-6-di-n-butylamino-fluoran (Compound No. W)

Te 100 g of 95% sulfuric acid, 12.2 g of 2-hydroxy4-di-n-butylamino-2'-carboxy-benzophenone and 6.1 g of
N-(4-methylphenyl)-2,4-diethoxyaniline were added and
reacted in the same manner as in Example 16. The product
swas recrystallized, whereby 11.4 g of slightly pink 2-(4-methylphenylamino)-3-ethoxy-6-di-n-butylamino-fluoran was
obtained. The melting point of this product was from 137
to 141°C. The product had λ<sub>max</sub> at 454 nm (MEC: 1.57 x
10<sup>4</sup>) and at 597 nm (MEC: 3.14 x 10<sup>4</sup>) as measured in 95%
acetic acid. A solution of this product in toluene was
colorless. When brought in contact with silica gel, the
product readily underwent color-development and turned
purplish black.

## Example 24

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2-(4-Methyphenylamino)-3-ethoxy-6-N-ethyl-N-ptolylamino-fluoran (Compound No. X)

To 100 g of 95% sulfuric acid, 12.3 g of 2-hydroxy-4-N-ethyl-N-p-tolylamino-2'-carboxy-benzophenone and 8.1 g of N-(4-methylphenyl)-2,4-diethoxyaniline were added and reacted in the same manner as in Example 16. The product was recrystallized from toluene, whereby 13.3 g of white 2-(4-methylphenylamino)-3-ethoxy-6-N-ethyl-N-p-tolylamino-fluoran was obtained. The melting point of this product was from 155 to  $158^{\circ}$ C. The product had  $\lambda_{max}$  at 459 nm (MEC: 1.84 x  $10^{4}$ ) and at 601 nm (MEC: 2.92 x  $10^{4}$ ) as measured in 95% acetic acid. A solution of this product in toluene was colorless. When brought in contact with

silica gel, the product readily underwent color-development and turned greenish black. Example 25

2-(4-Chlorophenylamino)-3-ethoxy-6-diethylaminofluoran (Compound No. Y)

To 100 g of 95% sulfuric acid, 9.4 g of 2-hydroxy-4-diethylamino-2'-carboxy-benzophenone and 8.8 g of N-(4-chlorophenyl)-2,4-diethoxyaniline were added and reacted in the same manner as in Example 16. The product was recrystallized, whereby 13.5 g of white 2-(4-chlorophenyl-amino)-3-ethoxy-6-diethylamino-fluoran was obtained. The melting point of this product was from 136 to 140°C. The product had λ<sub>max</sub> at 448 nm (MEC: 1.55 x 10<sup>4</sup>) and at 589 nm (MEC: 3.50 x 10<sup>4</sup>) as measured in 95% acetic acid. A solution of this product in toluene was colorless. When brought in contact with silica gel, the product readily underwent color-development and turned bluish black.

The N-(4-chlorophenyl)-2,4-diethoxyaniline used as a starting material was prepared in the same manner as in

Example 16 by using N-acetyl-2,4-diethoxyaniline and 4-bromochlorobenzene as the starting materials. The N-(4-chlorophenyl)-2,4-diethoxyaniline had a boiling point of from 188 to 190°C/0.5 mmHg.

Example 26

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2-(4-Chlorophenylamino)-3-ethoxy-6-di-n-butylaminofluoran (Compound No. Z)

To 100 g of 95% sulfuric acid, 11.1 g of 2-hydroxy-

4-di-n-butylamino-2'-carboxy-benzophenone and 8.8 g of N-(4-chlorophenyl)-2,4-diethoxyaniline were added and reacted in the same manner as in Example 16. The product was recrystallized, whereby 15.4 g of slightly pink 2-(4-5 chlorophenylamino)-3-ethoxy-6-di-n-butylamino-fluoran was obtained. The melting point of this product was from 116 The product had  $\lambda_{\text{max}}$  at 449 nm (MEC: 1.39 x  $10^4$ ) and at 590 nm (MEC: 3.48 x  $10^4$ ) as measured in 95% acetic acid. A solution of this product in toluene was colorless. When brought in contact with silica gel, the product readily underwent colordevelopment and turned bluish black.

Example 27

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2-(4-Chlorophenylamino)-3-ethoxy-6-N-ethyl-N-ptolylamino-fluoran (Compound No. AA) 15

To 100 g of 95% sulfuric acid, 12.3 g of 2-hydroxy-4-N-ethyl-N-p-tolylamino-2'-carboxy-benzophenone and 8.8 g of N-(4-chlorophenyl)-2,4-diethoxyaniline were added and reacted in the same manner as in Example 16. The product 20 was recrystallized from toluene, whereby 13.5 g of white 2-(4-cnlorophenylamino)-3-ethoxy-6-N-ethyl-N-p-tolylamino-fluoran was obtained. The melting point of this product was from 144 to 147.5°C. The product had  $\lambda_{max}$  at 453 nm (MEC: 1.68  $\times$  10<sup>4</sup>) and at 593 nm (MEC: 3.33  $\times$  10<sup>4</sup>) as measured in 95% acetic acid. A solution of this product in toluene was colorless. When brought in contact with silica gel, the product readily underwent colordevelopment and turned bluish black.

# Example 28

2-(4-Chlorophenylamino)-3-methoxy-6-N-ethyl-N-p-tolylamino-fluoran (Compound No. AB)

To 100 g of 95% sulfuric acid, 12.4 g of 2-hydroxy
4-N-ethyl-N-p-tolylamino-2'-carboxy-benzophenone and 8.0 g of N-(4-chlorophenyl)-2,4-dimethoxyaniline were added and reacted in the same manner as in Example 1. The product was recrystallized from toluene, whereby 12.9 g of white 2-(4-chlorophenylamino)-3-methoxy-6-N-ethyl-N-p-tolyl
amino-fluoran was obtained. The melting point of this product was from 209 to 213°C. The product had λ max at 454 nm (MEC: 1.66 x 10<sup>4</sup>) and at 591 nm (MEC: 3.13 x 10<sup>4</sup>) as measured in 95% acetic acid. A solution of this product in toluene was colorless. When brought in contact with silica gel, the product readily underwent color-development and turned bluish black.

Application Example 1

To 2.0 g of Compound A prepared in Example 1, 20 g of water and 20 g of an aqueous solution containing 10% by

20 weight of polyvinyl alcohol were added. The mixture was slowly dispersed and mixed in a ball mill at room temperature for 24 hours, whereby a colorless slurry was obtained wherein the particle size of the compound was about 3 μm. On the other hand, 7 g of bisphenol A was

25 added to 10 g of water and 40 g of an aqueous solution containing 10% by weight of polyvinyl alcohol. The mixture was slowly dispersed and mixed in a ball mill at

room temperature for 24 hours, whereby a slurry was obtained. The solid substance in the slurry had an average particle size of about 5 µm. Both slurries were mixed, and the mixture was uniformly dispersed and mixed at room temperature for 1 hour, whereby a slurry mixture was prepared. This slurry mixture was coated on one surface of a normal paper of 50 g/m<sup>2</sup> by means of a wire bar coater (wound wire: 0.35 mm in diameter) in an amount of the coated compound being 1.5 g/m<sup>2</sup> of the paper. The coated paper was dried in air at room temperature, whereby a heat sensitive record sheet having a substantially colorless heat sensitive layer was obtained. This heat sensitive record sheet will be referred to as No. a.

In the same manner, heat sensitive record sheets No.

15 b to No. o were prepared by using Compounds No. B to No.

O.

# Application Example 2

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To 2.0 g of Compound No. P prepared in Example 16, 20 g of water and 20 g of an aqueous solution containing 10% by weight of polyvinyl alcohol were added. The mixture was slowly dispersed and mixed in a ball mill at room temperature for 24 hours, whereby a colorless slurry was obtained wherein the particle size of the compound was about 3  $\mu$ m. On the other hand, 7 g of bisphenol A was added to 10 g of water and 40 g of an aqueous solution containing 10% by weight of polyvinyl alcohol. The mixture was slowly dispersed and mixed in a ball mill at room

temperature for 24 hour, whereby a slurry was obtained. The solid substance in the slurry had an average particle size of about 5 µm. Both slurries were mixed, and the mixture was uniformly dispersed and mixed at room

- temperature for 1 hour, whereby a slurry mixture was prepared. This slurry mixture was coated on one surface of a normal paper of 50 g/m<sup>2</sup> by means of a wire bar coater (wound wire: 0.35 mm in diameter) in an amount of the coated compound being 1.5 g/m<sup>2</sup> of the paper. The coated
- paper was dried in air at room temperature, whereby a heat sensitive record sheet having a substantially colorless heat sensitive layer was obtained. This heat sensitive record sheet will be referred to as No. p.

In the same manner, heat sensitive record sheets No. 15 q to No. ab were prepared by using Compounds No. Q to No. AB.

Further, for purpose of comparison, heat sensitive record sheets No. ac, No. ad and No. ae were prepared in the same manner as above by using known 2-anilino-3
20 methyl-6-diethylamino-fluoran (Compound No. AC), 2anilino-3-methyl-6-N-methyl-N-cyclohexylamino-fluoran (Compound No. AD) and 2-(2-chlorophenylamino)-6-diethylamino-fluoran (Compound No. AE).

These heat sensitive record sheets were subjected to the following tests.

(1) Color-development performance test

Heat sensitive record sheets No. a to No. ab and

comparative sheets No. ac, No. ad and No. ae were heated at a temperature of 150°C for 5 seconds, whereby the developed color hue, the developed color density and the initial color density were measured by means of Macbeth reflex densitometer RD-514 model with a black filter (Wratten # 106).

(2) Color-development characteristic test

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Heat sensitive record sheets No. a to No. ab and comparative sheets No. ac, No. ad and No. ae were heated for color-development within the temparature range of from 70 to 160°C for 5 seconds, whereby the color density at each temperature was measured in the same manner as in the above test (1), and the color-development initiation temperature and the rising for the color-development were calculated from the relationship between the temperature and the color density.

The results of the above-mentioned color-development performance test (1) and the results of the measurement of the color-development values (2) are shown in the following Table.

Color-development performance and color-development characteristic values of heat sensitive record sheets

/	Heat sensitive	sitive	Color-development	opment	Color-development	nen t
/	record s	heet	performance		characteristic	
/			Developed	Developed	Color-	Rising for
/	Compound	Sheet	or	color	development	
/	No.	No.	hue	density	initiation	development
					temperature( <sup>O</sup> C)	<b>3</b>
	æ	ದ	Purplish black	1,23	78.5	6.3
	æ	<b>a</b>	Purplish black	1.20	83.2	, r.
	ပ	υ	Purplish black	1.18	90.5	) 1
	Ω	ъ	Bluish black	1.22	200	
	EЭ	æ	Bluish black	1.20		
	Œ,	41	Bluish black	1.16	73.5	7.7
	<del>ن</del>	ס	Bluish black	1.24	92.5	4.76
	æ	£	Black	1,21		. α.
	н		Black	1.18		. 4
Present	ņ	٠.	Greenish black	1.23	81.0	
Invention	×	ᆇ	Dark green	1.25	78.4	7.7
	ıı	~	Reddish purple	1.17	86.5	. A
	Σ	E	Reddish purple	1.15	6	
	z	c	purplish black	1.20	83.0	. 4
	0	0	Dark blue	1.19		· r.
	Ω,	Q,	Bluish black	1.20	75.5	7 (
	o	<u>Б</u>	Ф	1.19	81,5	0 0
	~	ы	Bluish black	1.21	93,3	. r.
	တ	το:	Bluish black	1.24	0.96	
	Ţ	Ų	Purplish black	1.24	84.5	7 · V

	Heat sen	Bitive	Color-development	pment	Color-development	ent
	: 0)	heet	performance		characteristic values	
/			00	Developed	Color-	Rising for
/		2 toods	Developed	color	development	color-
/		No.	hue	density	Initiation	development
_		1			temperature(C)	
	13	11	Bluish black	1.17	78.0	4.9
	> >		Greenish black	1.16	76.5	7.7
	> 5	> 3	Section black	1.25	93.2	4.8
	3	₹	יייייייייייייייייייייייייייייייייייייי		ע	עע
present	×	×	Greenish black	17.1	C.10	) ) )
Tricontion	>	>	Bluish black	1.20	0.98	4.5
110 611 6 7 011	1 6	7 N		1.23	91.0	6.8
	4	י מי מי		1.17	79.7	4.1
		, u		1.23	94.5	3.9
		2 6	1.0	1.12	85	1.2
Comparative		ر د د			•	~ ~
Examples		ad	Reddish black	7.10	76	r :
	AE	ae	Reddish black	1.06	125	7.7
	1 0 000					

at a heating temperature of from 70 to  $160^{\circ}\mathrm{C}$  for a heating time of 5 seconds The color development by heating was conducted by means of lodiaceta model thermotester (manufactured by French National Fiber Research Institute) under a load of  $100 \text{ g/cm}^2$ . Note:

It is evident from the results shown in the above
Table that the heat sensitive record sheets wherein the
fluoran compounds of the present invention are used, are
far superior in the color-development performance and the
developed color characteristics to the comparative heat
sensitive record sheets wherein comparative fluoran
compounds are used. Thus, the industrial value for
practical application of the present invention is
considerably high.

#### CLAIMS:

A fluoran compound represented by general formula:

$$\begin{array}{c}
R_1 \\
R_2
\end{array}$$

$$\begin{array}{c}
OR_4 \\
NH-R_3
\end{array}$$
[1]

where each of  $R_1$  and  $R_2$  is a  $C_1$ - $C_4$  alkyl group, a cycloalkyl group or an aryl group, or  $R_1$  and  $R_2$  may form together with N a saturated ring,  $R_3$  is a benzyl or a phenyl group which may be substituted, and  $R_4$  is a  $C_1$ - $C_4$  alkyl group.

- 2. The fluoran compound according to claim 1, wherein  $R_4$  is a methyl group or an ethyl group.
- 3. The fluoran compound according to claim 2, wherein each of R<sub>1</sub> and R<sub>2</sub> is an ethyl group, a propyl group, a butyl group, a cyclohexyl group, a phenyl group which may be substituted by a methyl group, an ethyl group or a halogen atom, and R<sub>3</sub> is a benzyl group, a phenyl group or a halogen atom.
  - 4. The fluoran compound according to claim 1, which is 2-anilino-3-methoxy-6-di-n-butylamino-fluoran.
- 5. The fluoran compound according to claim 1, which is 20 2-anilino-3-methoxy-6-N-ethyl-N-p-tolylamino-fluoran.
  - 6. The fluoran compound according to claim 1, which is

2-(2-chlorophenylamino)-3-methoxy-6-di-n-butylamino-fluoran

- 7. The fluoran compound according to claim 1, which is 2-anilino-3-ethoxy-6-di-N-butylamino-fluoran.
- 5 8. The fluoran compound according to claim 1, which is 2-anilino-3-ethoxy-6-N-methyl-N-cyclohexylamino-fluoran.
  - 9. The fluoran compound according to claim 1, which is 2-anilino-3-ethoxy-6-N-ethyl-N-p-tolylamino-fluoran.
- 10. A heat sensitive record sheet which comprises a

  10 coated layer comprising a fluoran compound represented by
  the general formula:

$$R_1$$
 $R_2$ 
 $O$ 
 $OR_4$ 
 $NH-R_3$ 

where each of  $R_1$  and  $R_2$  is a  $C_1$ - $C_4$  alkyl group, a cycloalkyl group or an aryl group, or  $R_1$  and  $R_2$  may form together with N a saturated ring,  $R_3$  is a benzyl or phenyl group which may be substituted, and  $R_4$  is a  $C_1$ - $C_4$  alkyl group.

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Fluoran compounds.

fluoran compounds represented by general formula:

where each of R<sub>1</sub> and R<sub>2</sub> is an alkyl group, a cycloalkyl group or an aryl group, or R<sub>1</sub> and R<sub>2</sub> may form together with N a saturated ring, R<sub>3</sub> is a benzyl or phenyl group which may be substituted, and R<sub>4</sub> is an alkyl group, useful as color precursors for heat sensitive record sheets or electrical heat sensitive record sheets.

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# **EUROPEAN SEARCH REPORT**

0 1 Opplication Suffer

EP 83 10 9287

	DOCUMENTS CONS	IDERED TO BE RELEVA	NT	
Category		h indication, where appropriate, ant passages	Relevant to claim	CLASSIFICATION OF THE APPLICATION (Int. Cl. 3)
х	EP-A-0 005 379 * Claim 5; page	(DYNACHEM) = 47, example E *	1-3	C 07 D 493/10 B 41 M 5/26 C 07 D 493/10 C 07 D 311/00
х	FR-A-2 230 632 * Claims 1,3 *	(SHIN NISSO)	1,10	C 07 D 307/00
Y	GB-A-2 047 728	(FUJI PHOTO) = 7, example 12 *	1,10	
Y	GB-A-1 192 938	<del></del>	1,10	
P,X	EP-A-0 081 228 * Claim 1 *	(JUJO)	1,10	TECHNICAL FIELDS SEARCHED (Int. CI. <sup>3</sup> )
	·			C 07 D 493/00 B 41 M 5/00
<u>L</u>	The present search report has b	een drawn up for all claims		
Place of search  THE HAGUE  Date of completion of the search  30-01-1985			Examiner FARO I.	
Y: part doc A: tech O: non	CATEGORY OF CITED DOCL ticularly relevant if taken alone ticularly relevant if combined w ument of the same category nological background -written disclosure rmediate document	E : earlier   after th ith another D : docum L : docum	patent document e filing date ent cited in the a ent cited for othe er of the same pat	erlying the invention t, but published on, or pplication er reasons tent family, corresponding